

MICROCOPY RESOLUTION TEST CHART NATIONAL BUREAU OF STANDARDS 1463 A





OFFICE OF NAVAL RESEARCH

Contract N0014-80-C-0538

Task No. NR 051-736

TECHNICAL REPORT NO. 11

MATRIX CALIBRATION FOR THE QUANTITATIVE ANALYSIS

OF LAYERED SEMICONDUCTORS

BY SECONDARY ION MASS SPECTROMETRY

by

Alan A. Galuska

George H. Morrison*

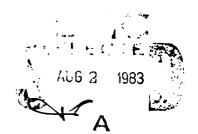
Prepared for Publication

in

Analytical Chemistry

Cornell University
Department of Chemistry
Ithaca, N. Y. 14853

July 27, 1983



Reproduction in whole or in part is permitted for any purpose of the United States Government

This document has been approved for public release and sale; its distribution is unlimited

83 08 01 •18

	REPORT DOCUMENTATION	PAGE	READ INSTRUCTIONS BEFORE COMPLETING FORM
	Technical Report No. 10	AISO 9 %	3. RECIPIENT'S CATALOG NUMBER
4.	TITLE (and Subtitle)		5. TYPE OF REPORT & PERIOD COVERED
	MATRIX CALIBRATION FOR THE QUANTITAL LAYERED SEMICONDUCTORS BY SECONDARY		Interim Technical Report
	SPECTROMETRY		6. PERFORMING ORG, REPORT NUMBER
7.	AUTHOR(a)		8. CONTRACT OR GRANT NUMBER(4)
	A. A. Galuska and G. H. Morrison		N00014-80-C-0538
9.	PERFORMING ORGANIZATION NAME AND ADDRESS		10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS
	Department of Chemistry Cornell University, Ithaca, N. Y.	14853	NR051-736
11.	CONTROLLING OFFICE NAME AND ADDRESS		12. REPORT DATE
1	ONR (472)	ļ	July 27, 1983
	800 N. Quincy St., Arlington, VA 2	22217	
<u> </u>	MONITORING AGENCY NAME & ADDRESS(II different		28 pages 15. SECURITY CLASS. (of this report)
'`	MONTONING AGENCY A AME & ACORESON BRIDGE		unclassified
			154. DECLASSIFICATION/DOWNGRADING SCHEDULE
16.	DISTRIBUTION STATEMENT (of this Report)		
	Approved for public release: dist	ribution unlimited	i
17.	DISTRIBUTION STATEMENT (of the abstract entered I	n Block 20, if different from	n Report)
18.	SUPPLEMENTARY NOTES		
	Prepared for publication in ANALYT		
19,	SIMS, superlattice, multilayer-multipractical ion yields, relative sense molecular beam epitaxy, ion implant	timatrix samples, sitivity factor, r	relative ion yields, Al_Ga,As,
sho sam pra tio	Analyses of Al Gal As matrices by some that secondary ion yields and specific composition. Calibration lines ctical ion yields, relative sensition lines formed using relative ion yield and relative sputter determination of a lib+ implant in	secondary ion mass uttering yields ar for Be, Si, B, P, vity factors, and ields provided sup	, and As were obtained using relative ion yields. Calibra- perior precision and accuracy.

MATRIX CALIBRATION FOR THE QUANTITATIVE ANALYSIS OF LAYERED SEMICONDUCTORS BY SECONDARY ION MASS SPECTROMETRY

A. A. Galuska and G. H. Morrison*

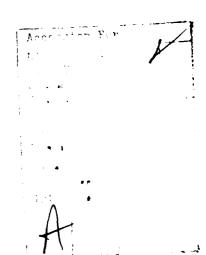
Department of Chemistry Cornell University Ithaca, New York 14853

BRIEF

The linearity of secondary ion yields and sputtering yields with composition was shown for ${\rm Al}_{\bf x} {\rm Ga}_{1-{\bf x}} {\rm As}$. Calibration lines for Be, Si, B, P, and As were obtained using practical ion yields, relative sensitivity factors and relative ion yields.

AC8303825





The sputtering and ionization of surface atoms by ion bombardment is the basis of secondary ion mass spectrometry (SIMS). High sensitivity for most elements and excellent depth resolution (less than 100 Å) have made SIMS an attractive method for concentration depth profiling. Although the technique had been applied successfully to a variety of materials, the complexity of the sputtering event has made quantitative analysis difficult. Secondary ion yields vary over several orders of magnitude. In addition, the ion yield of each element is influenced by a variety of factors including the nature of the primary ion beam (1-3), residual gas pressure (4), and matrix effects (5, 6).

The fabrication of external and internal standards by ion implantation has prodvided an accurate (± 15%) means of quantifying trace element distributions in homogeneous matrices (7-9). However, due to varying matrix effects, the quantitative analysis of heterogeneous matrices is extremely difficult. A depth profile of a sample composed of more than one matrix (in particular, layers of different matrices stacked one on top of the other) becomes distorted by the variation of secondary ion yields and sputtering yields with matrix: signal and time are no longer proportional to concentration and depth, respectively.

There are two principal explanations for the variation of secondary ion yields with matrix. It has been suggested that matrix effects are merely a function of sputtering yield (10, 11). Lower

sputtering yields enhance the build-up of reactive primary ions (0⁺ or Cs⁺) in surface layers resulting in increased secondary ion yields. Alternatively, others (12, 13) have asserted that ion yields are a linear function of matrix composition. Neither theory attempts to explain the variation of sputtering yields with matrix. Neither theory has been rigorously evaluated or applied.

Current trends in semiconductor technology require development of an applicable method of matrix calibration. Devices composed of Group III and V compound semiconductors (such as FET's, CW lasers, power FET's, IMPATT diodes, varactor diodes, and mixer diodes) are being rapidly developed (14-16) using growth techniques such as molecular beam epitary (MBE). These compound matrices are also being used to construct superlattices, alternating thin semiconductor layers of varying composition, Figure 1. There is a need for trace and major element quantification in and through these layers. The use of SIMS to monitor elemental distributions through thin layers and interfaces of varying composition is extremely difficult due to the changing matrix effects which are encountered. Proper calibration of ion yields and sputtering yields, however, can make quantitative concentration and depth measurements possible for these samples.

In this investigation, $\mathrm{Al}_{\mathbf{x}}\mathrm{Ga}_{1-\mathbf{x}}\mathrm{As}$ matrices grown by MBE and subsequently doped by ion implantation were examined for possible matrix calibration. Secondary ion yields and sputtering yields were found to be linear function of matrix composition. Calibrations using

practical ion yields, relative sensitivity factors, and relative ion yields were compared for precision and accuracy. The relative ion yield calibrations were precise and accurate to within 15%. Similarly, relative sputtering yield calibrations were precise and accurate to within 10%. The quantitative analysis of an ¹¹B⁺ implant into a multilayer-multimatrix sample was performed using relative sputtering yield and relative ion yield calibrations to correct the depth and concentration scales respectively.

EXPERIMENTAL

Sample Preparation. The Al_xGa_{1-x}As matrices were layers grown by MBE on semi-insulating GaAs substrates. The value of x was varied from 0 to 0.37 while the total atomic density remained at 4.43 X 10²² atom/cm³ ± 0.5%. The composition of the layers was determined by photoluminescence spectroscopy and verified to an accuracy of better than 10% (17) using Rutherford backscattering spectroscopy (RBS). Prior to implantation all samples were cleaned with acetone and methanol. A small piece of each sample was then mounted on an aluminum disk, using silver paint, for ion implantation. A set of samples was implanted with ⁹Be⁺ and ²⁸Si⁺, and a second set was implanted with ¹¹B⁺ and ³¹P⁺.

Instrumentation. Al_xGa_{1-x}As layers wasre grown in a VARIAN MBE-360 machine (18). RBS measurements were carried out on a GENERAL IONEX Tandetron Model 4110A. Analyses were performed using a 2.5 MeV He⁺ ion beam and solid state detection at a 170° angle from the incident beam path. Ion implantation was performed using two different ion implanters. Each was equipped with a hot filament ion source, a magnet for mass separation, and quadrupoles for ion beam focusing. Table I lists the implantation parameters.

SIMS analysis was carried out with a CAMECA IMS-3f ion microanalyzer using an electron multiplier in the pulse counting mode for signal detection (19). The instrument is interfaced to a HEWLETT PACKARD 9845T microcomputer for control and data acquisition. A 1.0 μ A 0_2^+ primary beam was rastered over a 400 x 400 μ m area at an energy of 5.5 KeV. The signal area was apertured down to a circle 60 μ m in diameter, and positive secondary ions were monitored. All analyses were performed with a residual pressure of 2 X 10^{-8} torr, and an energy window of 130 eV. A multiple sample holder allowed several samples to be mounted simultaneously. The depths of the sputtered craters were measured using a TALYSTEP stylus device with a resolution of 50 Å.

Software. Programs for instrumental control, data analysis, and matrix correction were written in BASIC for the HP 9845T. Data collection was performed using five second integrations per mass for each point during the depth profiles.

Procedure. Following ion implantation, samples were depth profiled in groups of four using a multiple sample holder. Groups of samples, including in each case GaAs, were inserted simultaneously to insure near identical analysis conditions. After allowing the pressure in the sample chamber to reach a steady-state condition, the primary beam was focused to a spot of about 100 μm in diameter, and the proper mass settings were determined. Without manipulating any instrumental parameters, the samples were analyzed consecutively until at least three profiles of each sample had been completed. In addition to the implants, ⁷⁵As+ was monitored for matrix quantification.

RESULTS AND DISCUSSION

Data Analysis. After profiling the samples, there are several ways in which practical ion yields (τ) can be calculated (8). In this study, elemental sensitivity was assumed constant throughout the full range of the implants. The secondary ion signals were integrated, and the background signals subtracted. Practical ion yields were then obtained using the implant fluences F (atom/cm²), integrated signals I (total counts), average atomic concentration C (atom/cm³), and the area A (cm²) and depth D (cm) of analysis:

$$\tau_{implant} = ions/atoms \cdot k_i = I_i/F_i \cdot A$$
 (1)

$$\tau_{\text{matrix}} = \text{ions/atoms} \cdot k_{\text{m}} = I_{\text{m}}/C_{\text{m}} \cdot A \cdot D$$
 (2)

where k_i and k_m are dimensionless instrumental factors representing the probability of detecting ions from the implant and matrix respectively. τ 's however can change drastically between analyses. Secondary ion collection and instrumental transmission are influenced by operator adjustment and indeterminate instrumental fluctuations which occur during routine operation. To reduce this effect, it is common practice to use relative sensitivity factors (RSF), as given by Eqs. (3) and (4), in which the τ of an analyte is ratioed to that of a reference element of in the same matrix. A knowledge of the average concentration of the reference element (C_{ref}) over the range of the analysis is required.

$$RSF_{implant} = \tau_{i} \cdot k_{ref} / \tau_{ref} \cdot k_{i}$$

$$= I_{i} \cdot C_{ref} \cdot D / F_{i} \cdot I_{ref}$$
(3)

$$RSF_{\text{matrix}} = \tau_{\text{m}} \cdot k_{\text{ref}} / \tau_{\text{ref}} \cdot k_{\text{m}}$$

$$= I_{\text{m}} \cdot C_{\text{ref}} / C_{\text{m}} \cdot I_{\text{ref}}$$
(4)

The normalization procedure is designed to minimize the influence of instrumental variations.

In addition to τ 's and RSF's, relative ion yields (R τ) were also examined. R τ 's are defined, for these analyses, as the ratio of the ion yield (τ_x) of an element in the sample matrix (Al_xGa_{1-x}As) to the ion yield (τ_0) of the element in a standard matrix (GaAs) when both measurements are performed under identical analysis conditions. The R τ 's for implant and matrix elements are given by Eqs. (5) and (6) respectively.

$$R\tau_{implant} = \tau_x/\tau_0 = I_x/I_0 \tag{5}$$

$$R\tau_{\text{matrix}} = \tau_{x}/\tau_{0} = I_{x} \cdot C_{0} \cdot D_{0}/I_{0} \cdot C_{x} \cdot D_{x}$$
 (6)

Like RSF's, RT's are designed to minimize the influence of instrumental variations. However, RT's have several advantages over RSF's for matrix calibration. First, RT's do not mask ion yield information the way RSF's can: RT's are directly proportional to individual ion yields while RSF's are related to the ion yields of both the analyte and the reference elements. Moreover, since the instrumental factors of the analyte and the reference (k_a and k_{ref}) can vary independently, RSF's are influenced by instrumental variations to a greater extent than the RT's. In addition, RT's are

designed to normalize matrix effects while RSF's are not.

In addition to ion yields, sputtering yields were also examined as a function of matrix composition. Sputtering yields (S) are defined in terms of the erosion rate \tilde{z} (cm/sec), atomic density of the sample N (atom/cm³), and the primary-ion current density J (ions/cm²-sec).

$$S = secondary atoms/primary atoms = {}^{o}_{z} \cdot N/J$$
 (7)

However, measurements of S are inconvenient due to the imprecision with which J is measured. To overcome this difficulty, the sputtering yields of the sample matrices $(Al_xGa_{1-x}As)$ have been normalized to the sputtering yields of the standard matrix (GaAs) measured under near identical conditions. Assuming J does not change between analyses, these relative sputtering yields (RS) take a simple form.

$$RS = S_x/S_0 = \overset{\circ}{z}_x \cdot N_x/\overset{\circ}{z}_0 \cdot N_0$$
 (8)

Since the atomic density of ${\rm Al}_{{\bf x}}{\rm Ga}_{1-{\bf x}}{\rm As}$ does not change substantially with x, Eq. (8) can be reduced to a ratio of erosion rates for these matrices. Consequently, relative sputtering yields provide a means of

determining the erosion rates of sample matrices regardless of current density.

<u>Precision and Accuracy</u>. If ion yields are linearly dependent on matrix composition, the practical ion yield of an element $g(\tau^g)$ can be described in the following manner (12):

$$\tau^{g} \propto \sum_{i=1}^{n} P_{i}^{g} \cdot C_{i}$$
 (9)

where n and C_i represent the total number of sample elements and the atomic concentration (atom/cm³) of element i, respectively. P_i^g is a dimensionless quantity representing the influence of element i on the ion yield of element g. Ignoring the effect of trace elements (elements of less than one atomic percent), Eq. (9) takes the following form for $Al_{\tau}Ga_{1-\tau}As$:

$$\tau^{g} = (P_{A1}^{g} - P_{Ga}^{g}) \cdot x + P_{Ga}^{g} + P_{As}^{g}.$$
 (10)

According to Eq. (10), a plot of \upsilon, RSF, or R\upsilon versus x should yield a straight line.

Tables II, III, and IV summarize the results obtained using v,

RSF, and Rr respectively. Each calibration is characterized by the date of analysis, standards used, y-intercept, slope, linear correlation, relative standard deviation of the slope, and the average relative standard deviation of a single point. In addition, analyses performed on different weeks were combined to form total calibrations. These total calibrations are useful to determine the precision of the calibration lines and the precision of individual points from week to week.

The quality of these calibration procedures can be evaluated in terms of the precision of the line, and the precision of each point. The Rt calibrations have excellent linearity followed by the t calibrations, and then the RSF calibrations. Similarly, the relative standard deviations of the slopes are smallest ($\simeq 9\%$) when Rt's are being used. In addition, the Rt calibration lines are reproducible from week to week while the t and RSF lines are not. The poorer precision of the t calibration lines can be attributed to the poor reproducibility of individual t measurements. The inferior quality of the RSF calibrations lines indicates that the ion yields of the analyte and the reference elements are not influenced proportionally by the matrix and instrumental effects which are encountered. Consequently, RSF's are not suitable for this type of calibration. Alternatively, Rt's are well suited to matrix calibration.

The quality of a calibration procedure can also be judged by the precision of each measurement. When measurements are limited to a

single analysis period, Rr's and RSF's give similar precisions (7% and 10% respectively) while t's are less precise (18%). However, when measurements are performed on different weeks, Rr's maintain good precision (9%) while the precision of the RSF's (24%) and t's (60%) deteriorates. The precision of RSF's is limited by those factors that influence the analyte and reference differently. Alternatively, the precision of Rr's is limited by the differences in the conditions under which the standard matrix and the sample matrix are analyzed. The limitations on the precision of the Rr's are more efficiently minimized than those of the RSF's. Overall, the Rr calibrations provide superior linearity and precision. Moreover, preliminary analysis indicates that an accuracy of 15% can generally be obtained using Rr calibrations.

In addition to relative ion yields, relative sputtering yields were also monitored as a function of matrix composition. As shown in Table V, the relationship is linear. The lines are quite precise and reproducible from week to week. Moreover, these RS calibration lines have been used to determine the erosion rates of $Al_xGa_{1-x}As$ matrices with an accuracy of better than 10%. Using RS calibrations to determine erosion rates and Rr calibrations to determine concentrations, it is now possible to quantitatively analyze multilayer-multimatrix samples.

Application. The use of Rt and RS calibration lines for quantitative analysis is straight forward. The concentration of

analyte at each point of a depth profile C_p (atom/cm³) can be determined using Eq. (11).

$$C_{p} = I_{p}/R\tau_{x} \cdot \tau_{0} \cdot D_{p} \cdot A \tag{11}$$

 $R\tau_x$ is the relative sputtering yield determined from a calibration line for the appropriate value of x, and τ_0 is the practical ion yield of the analyte in the standard matrix (GaAs). Ip is the signal and D_p is the depth increment associated with each point of the depth profile. Similarly, the erosion rate at each point of a depth profile z_p^2 can be determined using Eq. (12).

$$\dot{z}_{p} = RS_{x} \cdot \dot{z}_{0} \cdot N_{0} / N_{x} \tag{12}$$

 RS_x is the relative sputtering yield determined from a calibration line for the appropriate value of x, and $\overset{\circ}{z}_0$ is the sputtering rate of the standard matrix (GaAs). N_0 and N_x are the atomic densities of the standard matrix and the sample respectively. These calibrations may be used to analyze both homogeneous and multilayer-multimatrix samples.

Since these calibration lines are reproducible from week to week,

they are quite convenient for the quantitative analysis of homogeneous matrices. There are however two requirements. The sample and the standard matrices must be analyzed under identical conditions; consequently, a multiple sample holder is desireable. In addition, the matrix composition must be determined. This composition can be evaluated prior to SIMS analysis using a nondestructive technique such as RBS or photoluminescence. Alternatively, one may evaluate the matrix composition during the SIMS analysis. Since the concentration of As is constant for Al_xGa_{1-x}As matrices, the slope of the As calibration line and the Rt of As determined from the sample can be used to obtain the value of x.

$$x = (Rr - 1)/slope (13)$$

The accuracy of the SIMS and RBS measurements are comparable (15% and 10% respectively).

A multilayer-multimatrix sample can be quantitatively analyzed as a series of homogeneous matrices. Each layer may be treated as a homogeneous matrix. In addition, the interfaces, in which the composition changes from one matrix to another, can be approximated by linear concentration gradients. Alternatively, one may also treat each point of a depth profile as a homogeneous matrix by performing a point-by-point matrix calibration (A detailed description of this

process will be presented in a future paper.).

11_B+ KeV implant into (GaAs/Al 12 Ga 88 As/GaAs) provides a good example for the correction of matrix effects. In Figure 2a, the uncorrected SIMS depth profile of 11B+ (dashed line) and 75As+ (solid line) in this sample is presented. The first interface occurs at 33 time units while the second interface has not been reached in this profile. Due to matrix effects, the vertical and horizontal scales are no longer linearly related to concentration and depth, respectively. The peak in the 11B+ profile at 34 time units is an excellent example of a distortion introduce by the changing matrix effects at the interface. In Figure 2b, the horizontal scale has been transformed into depth using Eq. (12), and the concentration of AL (solid line) at each point has been calculated from the $^{75}\text{As}^+$ profile using Eq. (13). The $^{11}\text{B}^+$ profile was then corrected and quantified using Eq. (11). An Al concentration of 2.4 X 10^{21} atom/cm³ ± 6.0% was calculated for the $\mathrm{Al_xGa_{1-x}As}$ layer which agrees to within experimental error with the value obtained from the RBS analysis $(2.7 \times 10^{21} \text{ atom/cm}^3)$. The correction procedure had also removed the distorted interface region of the 11B+ profile, transforming the profile into the expected gaussian shape. Thus, the application of this correction procedure can make a tremendous difference in the evaluation of SIMS profiles through layered multimatrix structures.

ACKNOWLEDGMENT

The authors gratefully acknowledge the assistance of C. Palmstrom and J. Mayer with the RBS measurements, and B. Shaft for the growth of the MBE matrices. Also acknowledged is the help of H. Dietrich of the Naval Research Laboratories (NRL) for his assistance with the ion implants. The ion implants were performed at both NRL and the National Research and Resource Facility for Submicron Structures at Cornell.

CREDIT

This work was supported by the National Science Foundation and the Office of Naval Research.

LITERATURE CITED

- (1) Ishitani, T.; Tamara, H.; Shinmiyo, T. <u>Surf. Sci.</u> 1976, 55, 179-188.
- (2) Storms, H. A.; Brown, K. F.; Stein, J. D. Anal, Chem. 1977, 49, 2023-2030.
- Barcz, A.; Domanski, M.; Woitowicz-Natanson, B. "Secondary Ion Mass Spectrometry SIMS III"; Benninghoven, A.; Giber, J.; Riedel, M.; Werner, H. W., Eds.; Springer-Verlag: New York, 1981, 134-139.
- (4) Morgan, A. E.; Werner, H. W. Anal. Chem. 1976, 48, 699-708.
- (5) McCracken, G. M. Rep. Prog. Phys. 1975, 38, 241-327.
- (6) Blaise, G.; Bernheim, M. <u>Surf. Sci.</u> 1975, 47, 324-343.
- (7) Gries, W. H. <u>Int.</u> <u>J. Mass Spectrum</u>. <u>Ion Phys.</u> 1979, 30, 97-112.
- (8) Leta, D. P.; Morrison, G. H. Anal. Chem. 1980, 52, 514-519.
- (9) Leta, D. P.; Morrison, G. H. Anal. Chem. 1980, 52, 277-280.
- (10) Deline, V. R.; Katz, W.; Evans, C. A.; Williams, P. Appl. Phys. Lett. 1978, 33, 832-834.
- (11) Deline, V. R.; Evans, C. A.; Williams, P. Appl. Phys. Lett. 1978, 33, 578-580.
- (12) Slodzian, G. "Secondary Ion Mass Spectrometry SIMS III"; Benninghoven, A.; Giber, J.; Laszlo, J.; Riedel, M.; Werner, H. W., Eds.; Springer-Verlag: New York, 1981, pp 115-123.
- (13) Steele, I.; Herrig, R.; Hutcheon, I. Proceedings of the 15th Annual Conference of the Microbeam Analysis Society, San Francisco, CA, 1980, pp 151-153.
- (14) Panish, M. B. Science 1980, 20, 916-922.
- (15) Cho, A. Y.; Arthur, J. R. "Progress in Solid State Chemistry"; Somorgai, G.; McCaldin, J., Eds.; Pergamon: New York, 1975; Vol. 10, pp 157-191.
- (16) Cho, A. Y. J. Vac. Sci. Technol. 1979, 16, 275-284.

- (17) Mayer, J. W.; Ziegler, J. F.; Chang, L. L.; Tsu, R.; Esaki, L. <u>J. Appl. Phys.</u> 1973, 44, 2322-2325.
- (18) Daries, G. J.; Heckingbottom, R.; Ohno, H.; Wood, C. E. C.; Calawa, A. R. Appl. Phys. Lett. 1980, 37, 290-292.
- (19) Ruberol, J. M.; Lepareur, M.; Autier, B.; Gourgout, J. M. VIIIth International Congress on X-ray Optics and Microanalysis and 12th Annual Conference of the Microbeam Analysis Society, Boston, MA, 1977, pp 133A-133D.

Table I. Implantation Parameters

	$A1_{x}Ga_{1-x}As$				
Implant Element	Matrix (x)	Fluence (atom/cm ²)	Energy (KeV)	Source	
9 Be	0-0.37	1 X 10 ¹⁴	2 50	. Be solid	
11 _B	0-0.37	1 X 10 ¹⁴	250	BF ₃ gas	
28 _{Si}	0-0.37	1 X 10 ¹⁵	2 50	SiF _{4 gas}	
31 _P	0-0.37	1 X 10 ¹⁵	300	PF ₂ gas	

Table II. Practical Ion Yield Calibrations for $Al_xGa_{1-x}As$ Matrices

Analyte	Week No.	X Values	Intercept X 10-7	Slope X 10 ⁻⁵	Linear Correlation	RSD Slope %
9 _{Be}	1	0,.12,.26,.37	1.23	41.5	0.927	32.8
	2	0,.13,.21,.37		67.3	0.915	68.6
	3	0,.13,.18,.21		32.3	0.995	9.5
²⁸ Si	1		36.4	22.2	0.989	18.5
	1 2		25.3	15.2	0.987	17.0
	3		38.9	17.7	0.984	10.8
31 _P	16	0,.18,.26,.31	0.822	0.291	0.936	18.6
	32	0,.13,.18,.37		0.258	1.000	2.0
11 _B	16		25.8	7.05	0.734	53.6
	32		28.2	17.0	0.994	8.7
75 _{As}	1		1.78	0.130	0.966	14.6
	2		0.932	0.0891	0.915	21.0
	3		0.940	0.108	0.984	11.0
	32		0.918	0.0532	0.932	11.2
		Avg RSD pe	r point =	18.4%		
9Be	Tota1		31.4	44.7	0.868	12.3
40Si	Tota1		11,0	19.0	0.994	5.8
31 _P	Tota1		1.00	0.278	0.928	12.4
11 _B	Tota1		27.0	15.3	0.825	18.8
75 As	Total		0.642	0.111	0.722	19.2

Avg RSD per point = 60.4%

Table III. Relative Sensitivity Factor Calibrations for ${\rm Al}_{\bf x}{\rm Ga}_{1-{\bf x}}{\rm As}$ Matrices

Analyte Reference	Week No.	X Values	Intercept	Slope	Linear Correlation r ²	RSD Slope
9 _{Be/75} As	1	0,.12,.26,.37	47,0	890	0.731	33.6
	2	0,.13,.21,.37	84,8	573	0.894	23.0
	3	0,.13,.18,.21	66.5	950	0.995	4.7
28 _{Si/75As}	1		39.8	271	0.708	38.2
	1 2		39,4	3 82	0.903	24.2
	3		36.0	711	0.936	37.8
31 _{P/} 75 _{As}	16	0,.18,.26,.31	1.58	9.67	0.715	40.3
1 / 113	32	0,.13,.18,.37	1.88	4.91	0.827	58.9
11 _{B/} 75 _{As}	16		30.7	454	0.945	26.1
DI AS	32		60.9	509	0.940	17.1
		Avg RSD per	r point = 1	10.4%		
9Be/75As	Total	•	88.7	740	0.694	16.8
28Si/75As	Total		42.7	357	0.804	17.7
31 _{P/} 75 _{As}	Total		2.41	4.45	0.464	79.2
11 _{B/75} As	Total		34.0	481	0.883	19.2

Avg RSD per point = 24.4%

Table IV. Relative Ion Yield Calibrations for ${\rm Al}_{\bf x}{\rm Ga}_{1-{\bf x}}{\rm As}$ Matrices

Analyte	Week No.	X Values	Intercept	S1 op e	Linear Correlation r ²	RSD Slope
9 _{Be}	1	0,.12,.26,.37	1.0	52.9	0,999	6.4
	2	0,.13,.21,.37	1.0	42.6	0.996	4.3
	.3	0,.13,.18,.21	1.0	41.7	1.000	14.2
²⁸ Si	1		1.0	44.5	1,000	9.8
	2		1.0	40.1	0.975	5.7
	3		1.0	45.9	1.000	9,6
31 _P	16	0,.18,.26,.31	1.0	21.2	0,995	9.6
•	32	0,.13,.18,.37	1.0	23.0	1.000	2.5
11 _B	16		1.0	46.7	0.990	10,7
_	32		1.0	39.5	1.000	8.6
75 _{As}	1	-	1.0	6.00	0.933	15.7
***	2		1.0	6.54	0.991	2.8
	3		1.0	5.76	0.980	12.4
	32		1.0	5.32	0.925	11.0
		Avg RSD per	point = 7.	4%		
9Be	Tota1		1.0	46.3	0.981	13.7
4051	Tota1		1.0	44.7	0.986	5.0
210	Tota1		1.0	22.1	0.942	4.6
11 _B	Tota1		1.0	46.6	0.961	7.0
75 As	Tota1		1.0	5.76	0.917	7.0

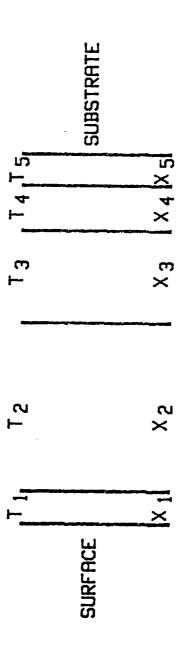
Avg RSD per point = 9.2%

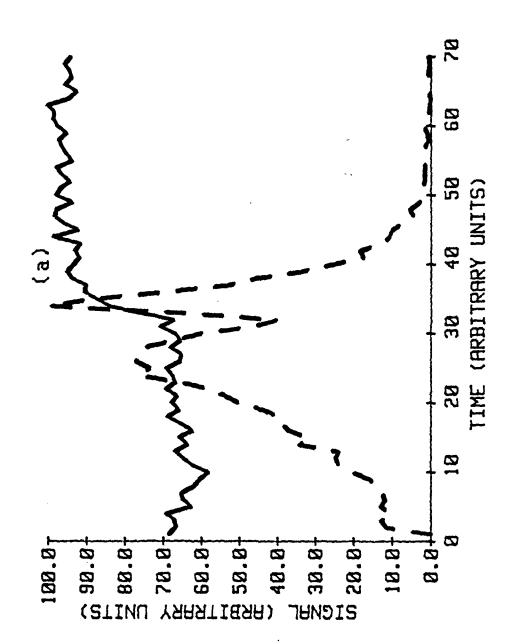
Table V. Relative Sputtering Yield Calibrations for ${\rm Al}_{\bf x}{\rm Ga}_{1-{\bf x}}{\rm As}$ Matrices

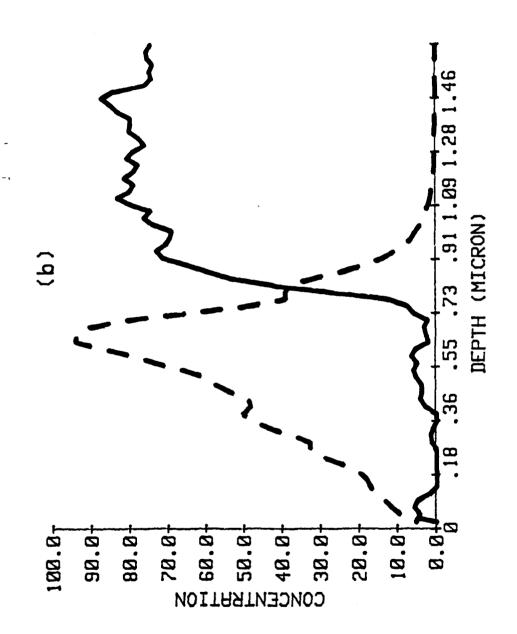
Week No.	Intercept	Slope	Linear Correlation r ²	RSD Slope %
1	1.0	-1.00	0.928	11.8
2	1.0	-1.01	0.913	12.5
3	1.0	-0.89	0.991	4.0
16	1.0	-0.88	0.997	3.2
32	1.0	-0.75	0.934	11.6
	Avg RSD pe	r point = 3.89	*	
Total	1.0	-0.86	0.906	8.3
	Avg RSD pe	r point = 7.99	5.	

FIGURE CAPTIONS

- Figure 1. A hypothetical ${\rm Al}_{\bf x}{\rm Ga}_{1-{\bf x}}{\rm As}$ superlattice. The thickness T of the layers can vary from several angstroms to several microns while x can be varied from 0 to 1.
- Figure 2. SIMS depth profile of a 250 Kev ¹¹B⁺ implant into a GaAs/Al_{0.12}Ga_{0.88}As/GaAs sample. (a) uncorrected profile of ¹¹B⁺ (---) and ⁷⁵As⁺ (---); (b) concentration profiles of B (---) {2.0 X 10¹⁸ atom/cm³ full scale } and Al (---) {1.0 X 10²² atom/cm³ full scale}.







TECHNICAL REPORT DISTRIBUTION LIST, JEN

	No. Copies		10. 12:18 s
Office of Naval Research		U.S. Army Research Office	
Attn: Code 472		Attn: CRD-AA-IP	
300 North Quincy Street		P.O. Box 1211	
Arlington, Virginia 22217	2	Research Triangle Park, N.C. 27709	:
ONR Branch Office		Naval Ocean Systems Center	
Attn: Dr. George Sandoz		Attn: Mr. Joe McCartney	
536 S. Clark Street	1	San Diego, California 92152	<u> </u>
Chicago, Illinois 60605	Ł	Naval Weapons Center //	
OMR Area Office		Attn: Dr. A. B. Amster,	
Attn: Scientific Dept.		Chemistry Division	
715 Broadway		China Lake, California 93555	1
New York, New York 10003	1	• • · · · · · · · · · · · · · · · ·	
,	~	Naval Civil Engineering Laboratory	
ONR Western Regional Office ?		Attn: Dr. R. W. Drisko	
1030 East Green Street		Port Hueneme, California 93401	:
Pasadena, California 91106	1		
		Department of Physics & Chemistry	
ONR Eastern/Central Regional Office		Naval Postgraduate School	
Attn: Or. L. H. Peebles		Montarey, California 93940	:
Building 114, Section D			
666 Summer Street	•	Dr. A. L. Slafkosky	
Boston, Massachusetts 02210	1	Scientific Advisor	
Discuss Namel Bassanch Inhanatama		Commandant of the Marine Corps (Code Rh-1)	
Director, Naval Research Laboratory Attn: Code 6100		(Code RD-1) Washington, D.C. 20380	•
Washington, D.C. 20390	1	wastixtigeou, p.o. 20000	•
washington, b.o. 20000	•	Office of Naval Research	
The Assistant Secretary		Attn: Dr. Richard S. Miller	
of the Navy (RE&S)		800 N. Quincy Street	
Department of the Navy		Arlington, Virginia 20217	•
Room 4E736, Pentagon			
Washington, D.C. 20330	1	Naval Ship Research and Development Center	
Commander, Naval Air Systems Command		Attn: Dr. G. Bosmajian, Applied	
Attn: Code 310C (H. Rosenwasser)		Chemistry Division	
Department of the Mavy		Annapolis, Maryland 21401	1
Washington, D.C. 20360	1	_	
		Naval Ocean Systems Center	
Defense Technical Information Center		Attn: Dr. S. Yamamoto, Marine	
Building 5, Cameron Station	12	Sciences Division	
Alexandria, Virginia 22314	12	San Diego, California 91232	•
Dr. Fred Saalfeld		Mr. John Boyle	
Chemistry Division, Code 6100		Materials Branch	
Maval Research Laboratory	_	Naval Ship Engineering Center	
Washington, D.C. 20375	1	Philadelphia, Pennsylvania 19112	
Dr. Rudolph J. Marcus	1	Mr.James Kelley	:
Office of Naval Research		DTNSRDC Code 2903	
Scientific Liaison Group - Amer. E	bassy	Annapolis, Maryland 21;02	
A.P.O. San Francisco, CA. 96503		·	

TECHNICAL REPORT DISTRIBUTION LIST, 0510

·	No. Copies		<u>No.</u> Copies
Dr. M. B. Denton		Dr. John Duffin	
Department of Chemistry		United States Naval Postgraduate	
University of Arizona		School	
Tueson, Arizona 85721	1	Monterey, California 93940	1
racion, arribina objet	•	noncerey, delitornia 15.40	•
Dr. R. A. Osteryoung		Dr. G. M. Hieftje	
Department of Chemistry		Department of Chemistry	
State University of New York		Indiana University	
at Buffalo		Bloomington, Indiana 47401	1
Buffalo, New York 14214	1	,	
··· ,		Dr. Victor L. Rehn	
Dr. B. R. Kowalski		Naval Weapons Center	
Department of Chemistry		Code 3813	
University of Washington		China Lake, California 93555	
Seattle, Washington 98105 .	1		
,		Dr. Christie G. Enke	
Dr. S. P. Perone		Michigan State University	
Department of Chemistry		Department of Chemistry	
Purdue University		East Lansing, Michigan 43824	•
Lafayette, Indiana 47907	1		
•		Dr. Kent Eisentraut, MBT	
Dr. D. L. Venezky		Air Force Materials Laboratory	
Naval Research Laboratory		Wright-Patterson AFB, Ohio 45433	:
Code 6130			
Washington, D.C. 20375.	I	Walter G. Cox, Code 3632	
-		Naval Underwater Systems Center	
Dr. H. Freiser		Building 148	
Department of Chemistry		Newport, Rhode Island 02340	1
University of Arizona	1	• •	
Tuscon, Arizona 85721	·	Professor Isiah M. Warner	
		Texas A&M University	
Or. Fred Saalfeld		Department of Chemistry	
Naval Research Laboratory		College Station, Texas 77840	1
Code 6110			
Washington, D.C. 20375	1	Professor George H. Morrison	
		Cornell Carversity	
Dr. H. Chernoff		Department of Chemisty	
Department of Mathematics •		Ithaca, New York 14853	÷
Massachusetts Institute of Technology		•	
Cambridge, Massachusetts 02139	1	Professor J. Janata	
		Department of Bioengineering	
Dr. K. Wilson		University of Utah	
Department of Chemistry		Salt Lake City, Utah 84112	1
University of California, San Diego			
La Jolla, California	1	Dr. Carl Heller	
		Naval Weapons Center	
Or. A. Zirino		China Lake, California 93555	1
Naval Undersea Center			
San Diego, California 92132	1	Dr. L. Jarvis	1
		Code 6100	
		Naval Fesearch Laboratory	
		Washington, D. C. 20375	

